

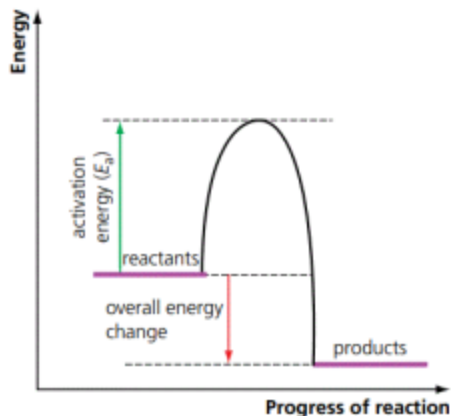
# Exothermic and Endothermic Reactions

**Exothermic**

Energy given OUT to the surroundings as heat

The temperature of the surroundings INCREASE

Example: Combustion reactions



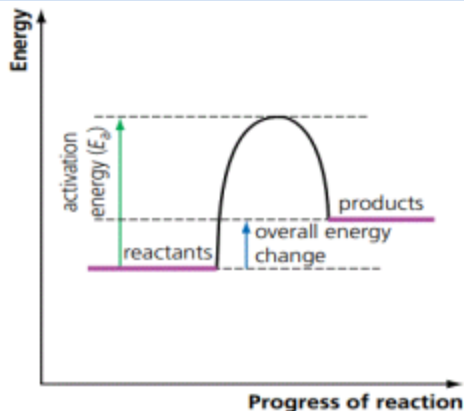
Profile

**Endothermic**

Energy is taken IN from the surroundings

The temperature of the surroundings DECREASE

Example: Thermal decomposition



Profile